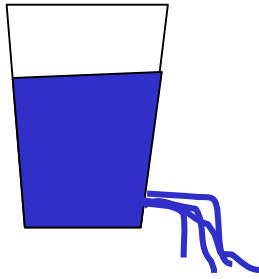


# **Electricity Merit Badge**

## **DC**

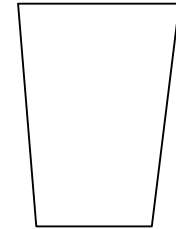
## **Direct Current**

Student Copy



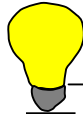
Glass of Water has Force called pressure

# Direct Current

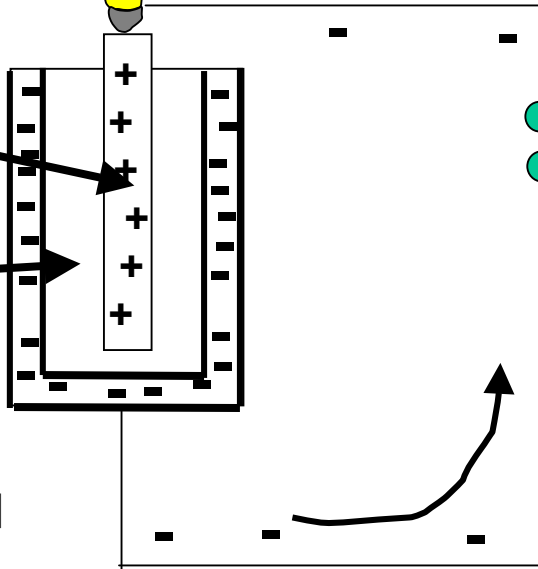


Empty Glass Has no Force

Flashlight



Carbon Rod  
Gel Insulator  
Zinc Casing

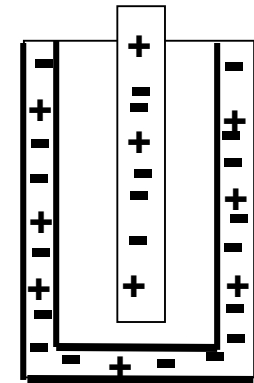


Switch

Battery works from a chemical reaction between the carbon rod and zinc case

**Charged Battery**  
**1.5 volts**

**Current Flow**    **Electrons**



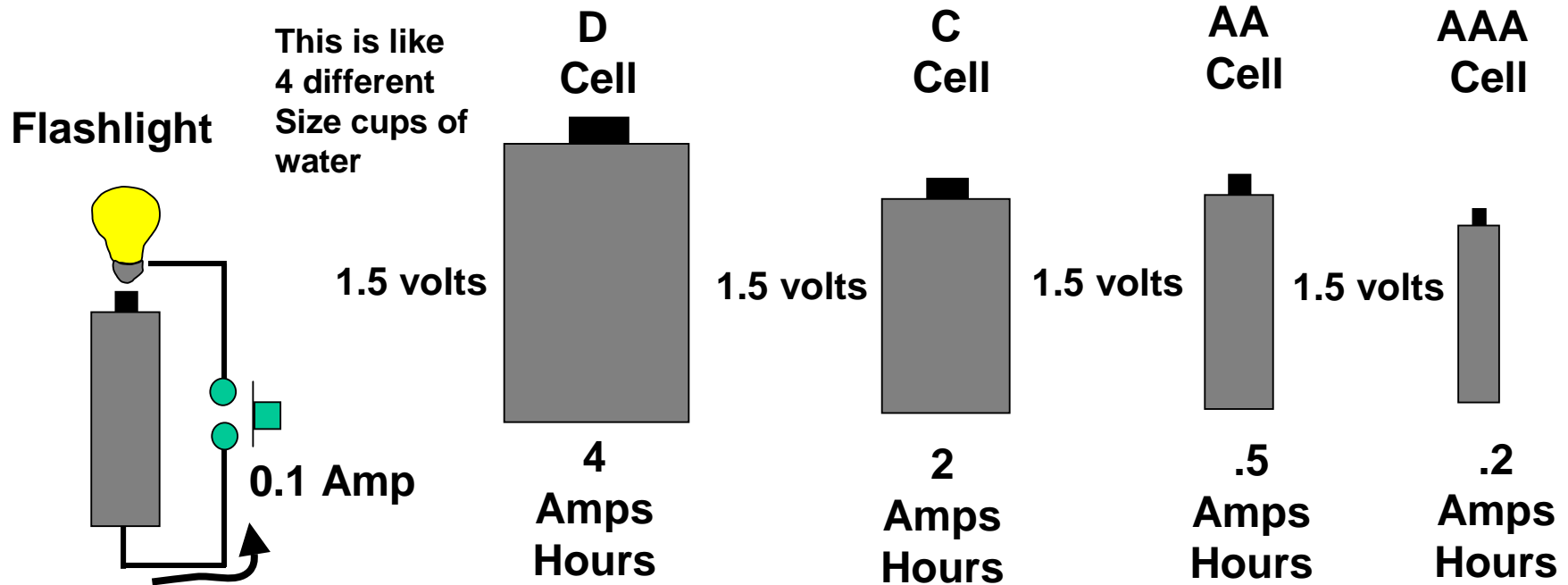
**Dis-Charged Battery**  
**No Voltage**

Voltage is the quantity of electrical force    Measured in Volts  
Current is the flow of electrons    Measured in Amps  
DC Stand for Direct Current  
DC is current flowing in one direction

# Direct Current

Pass out

## Typical Battery Rating - How Long will they Last



If a flashlight pulls .1 Amp, how long will each battery last?

Time = Amp Hours / Load in Amps

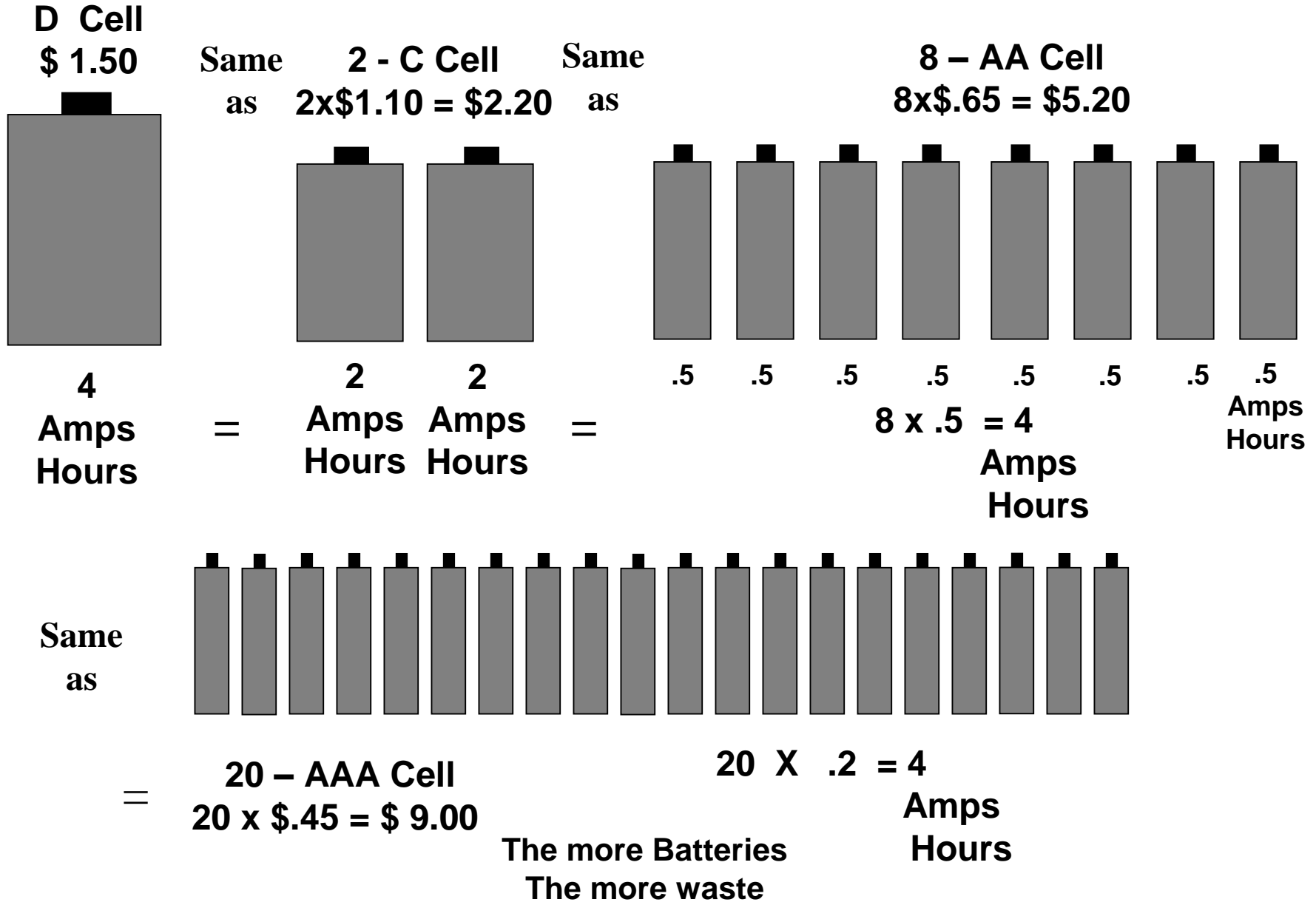
D cell =  $\frac{\text{Amp Hours}}{\text{Amps}}$  = \_\_\_\_\_ = \_\_\_\_\_ hours

C cell =  $\frac{\text{Amp Hours}}{\text{Amps}}$  = \_\_\_\_\_ = \_\_\_\_\_ hours

AA cell =  $\frac{\text{Amp Hours}}{\text{Amps}}$  = \_\_\_\_\_ = \_\_\_\_\_ hours

AAA cell =  $\frac{\text{Amp Hours}}{\text{Amps}}$  = \_\_\_\_\_ = \_\_\_\_\_ hours

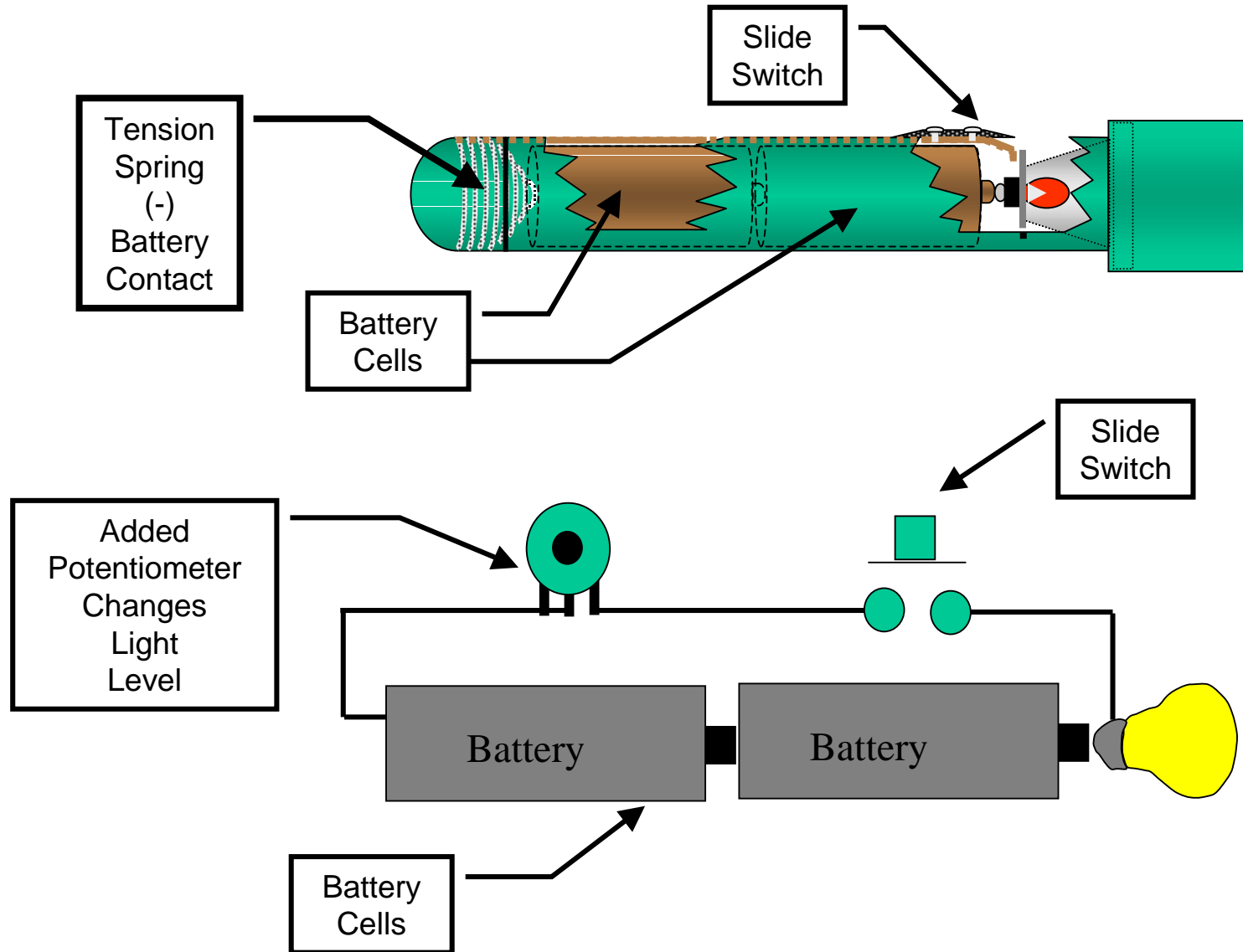
# Direct Current Cost of Batteries for the Same Output



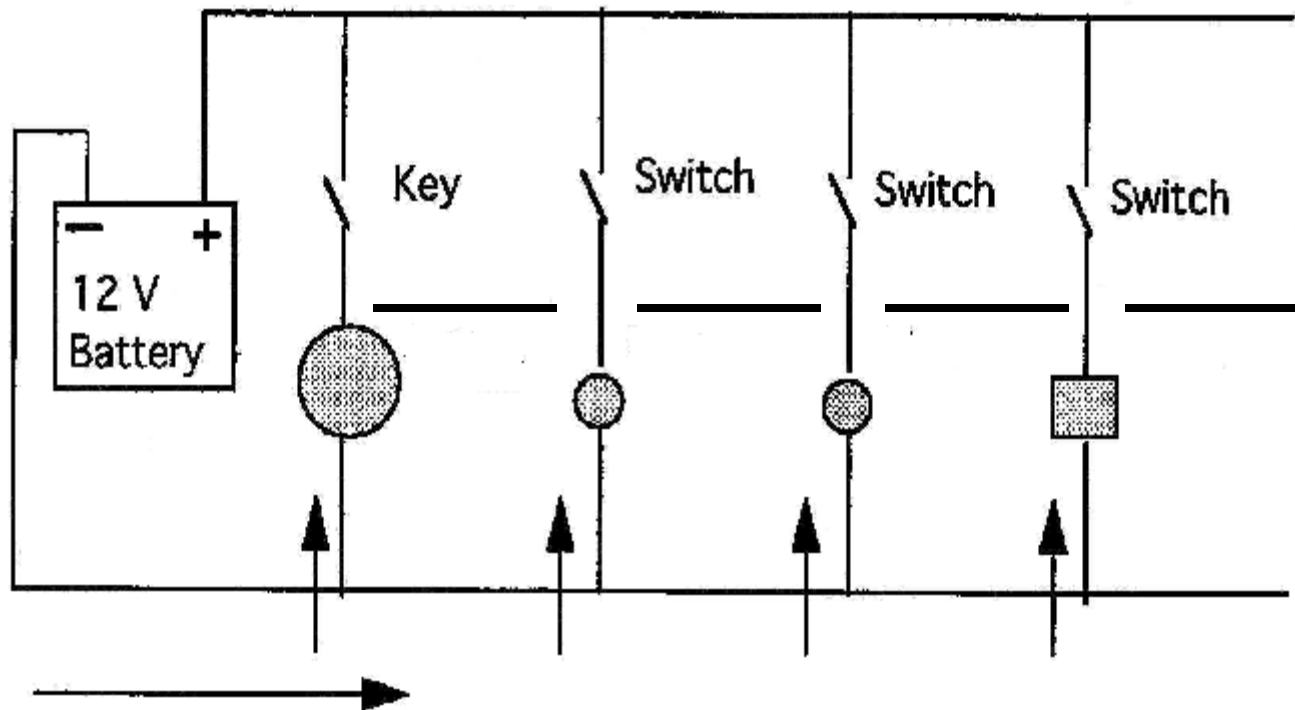
# Many Battery Types

- **Zinc-carbon battery** - Also known as a **standard carbon** battery, zinc-carbon chemistry is used in all inexpensive AA, C and D dry-cell batteries. The electrodes are zinc and carbon, with an acidic paste between them that serves as the electrolyte.
- **Alkaline battery** - Used in common Duracell and Energizer batteries, the electrodes are zinc and manganese-oxide, with an alkaline electrolyte.
- **Lithium photo battery** - Lithium, lithium-iodide and lead-iodide are used in cameras because of their ability to supply power surges
- **Lead-acid battery** - Used in automobiles, the electrodes are made of lead and lead-oxide with a strong acidic electrolyte (rechargeable).
- **Nickel-cadmium battery** - The electrodes are nickel-hydroxide and cadmium, with potassium-hydroxide as the electrolyte (rechargeable).
- **Nickel-metal hydride battery** - This battery is rapidly replacing nickel-cadmium because it does not suffer from the memory effect that nickel-cadmiums do (rechargeable).
- **Lithium-ion battery** - With a very good power-to-weight ratio, this is often found in high-end laptop and cell phones (rechargeable).
- **Zinc-air battery** - This battery is lightweight and rechargeable.
- **Zinc-mercury oxide battery** - This is often used in hearing-aids.
- **Silver-zinc battery** - This is used in aeronautical applications because the power-to-weight ratio is good.
- **Metal-chloride battery** - This is used in electric vehicles

# Flashlight Diagram

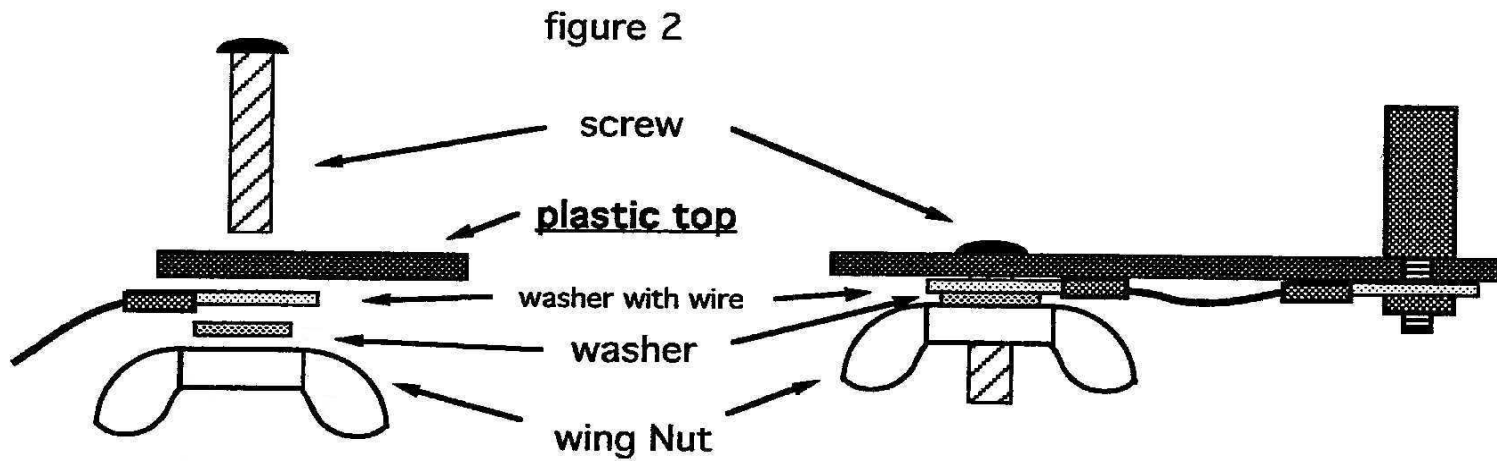
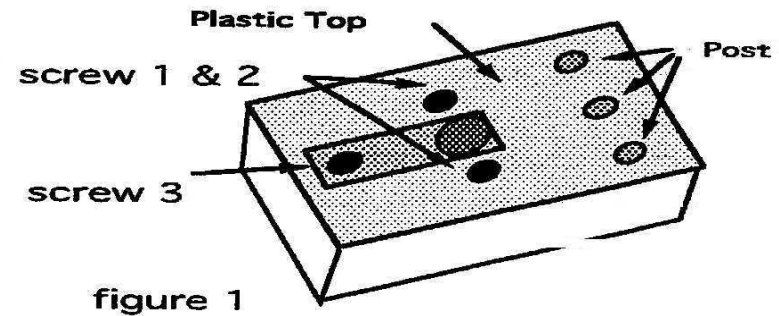
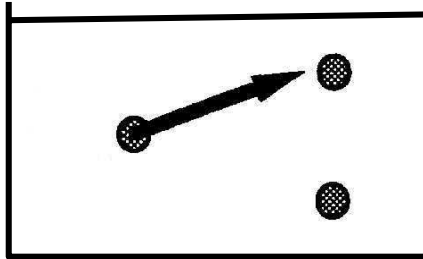


## Car DC Electrical System



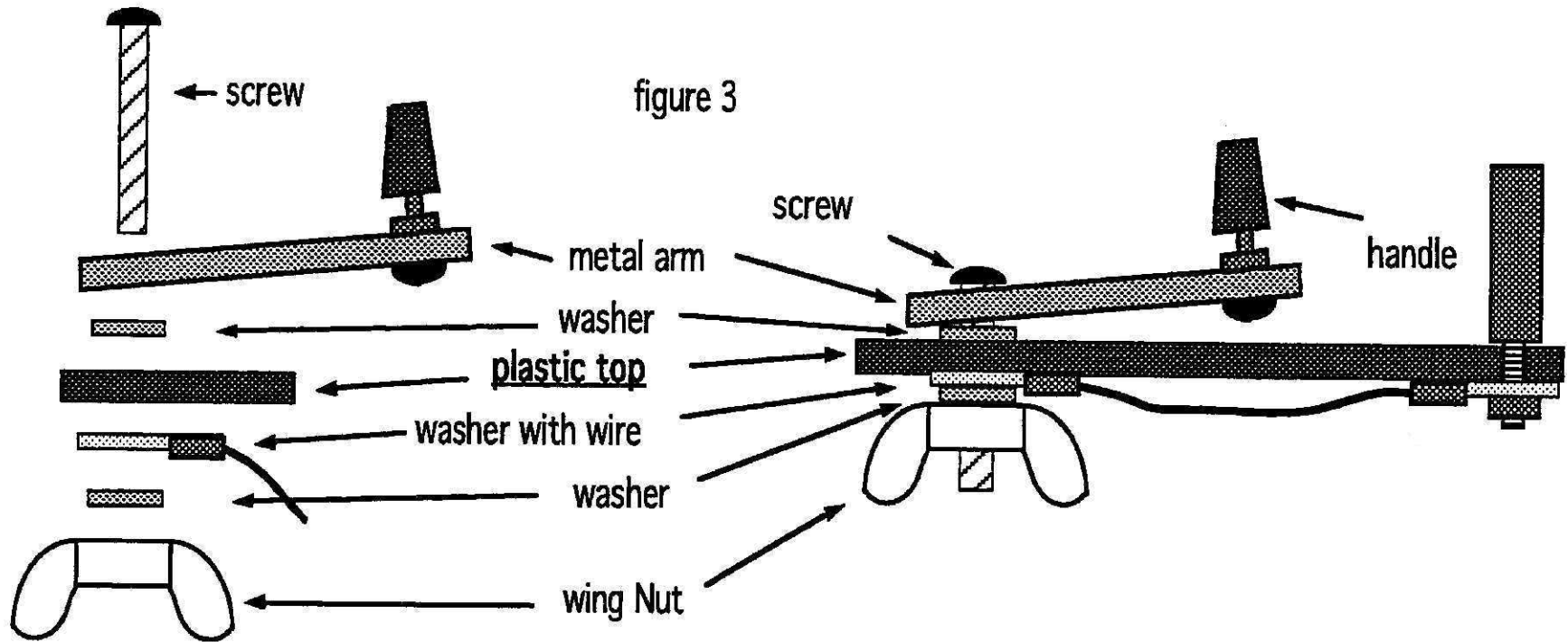
Name four electrical items in a car

# Build an Electrical Switch



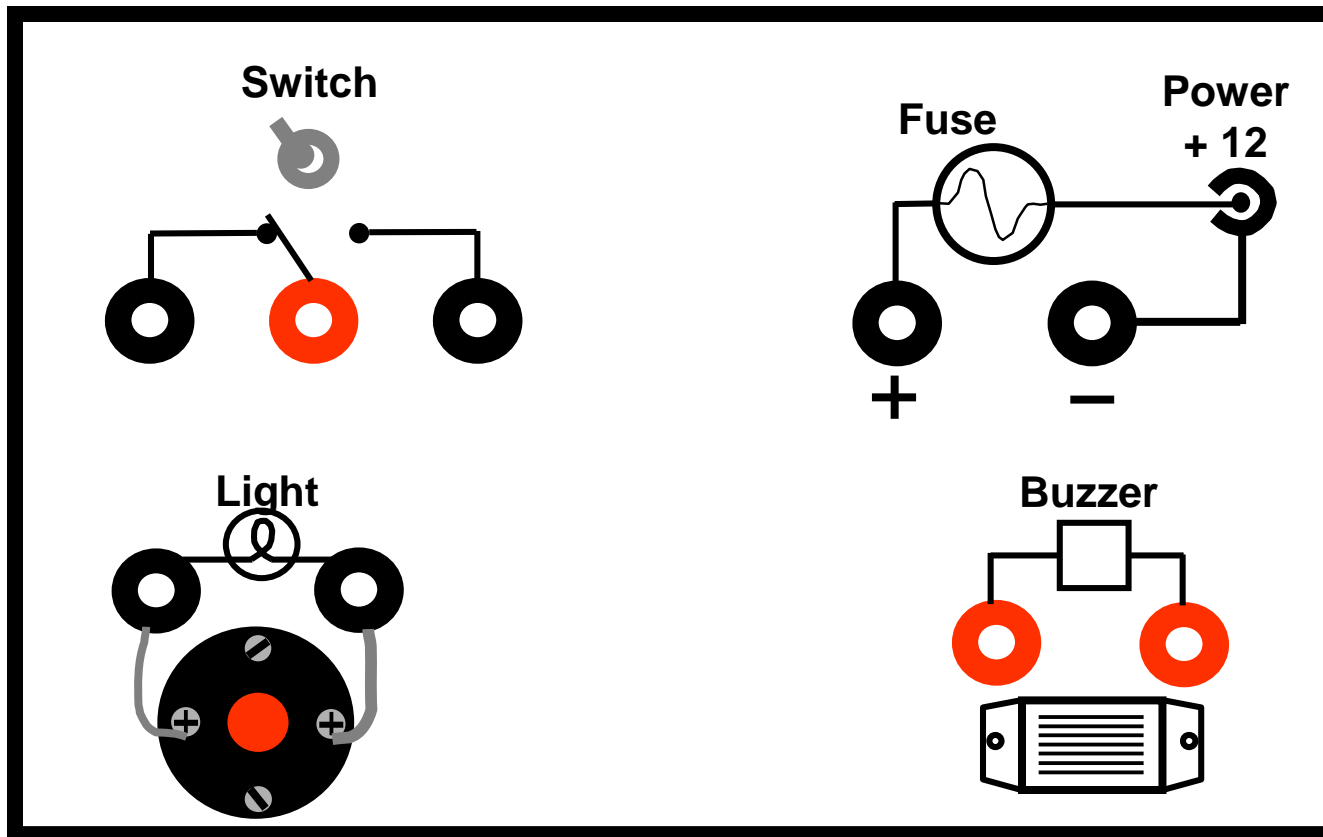


# Build an Electrical Switch



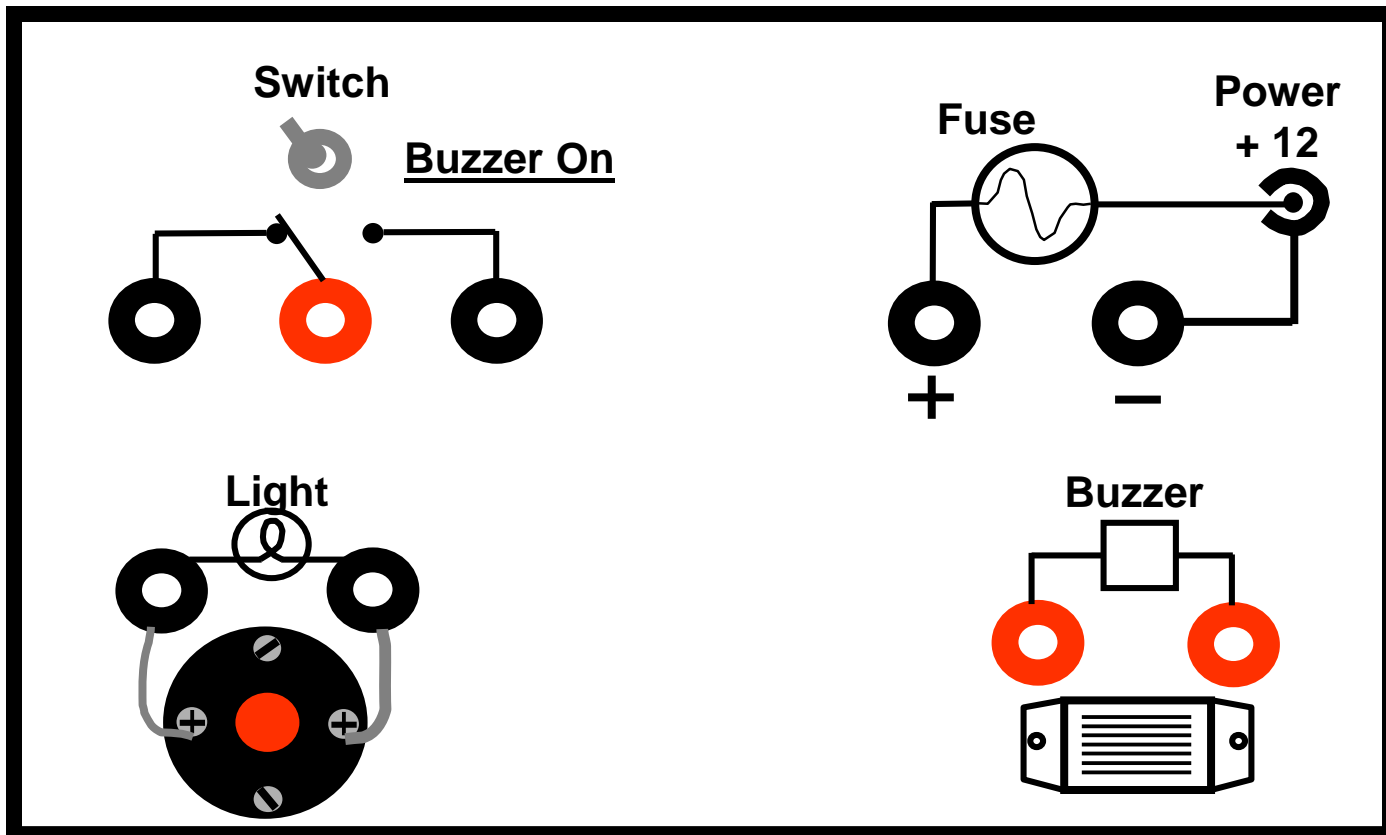
## Direct Current Test Box

Draw 4 different wiring test circuits, then pass out boxes.



# Direct Current Test Box

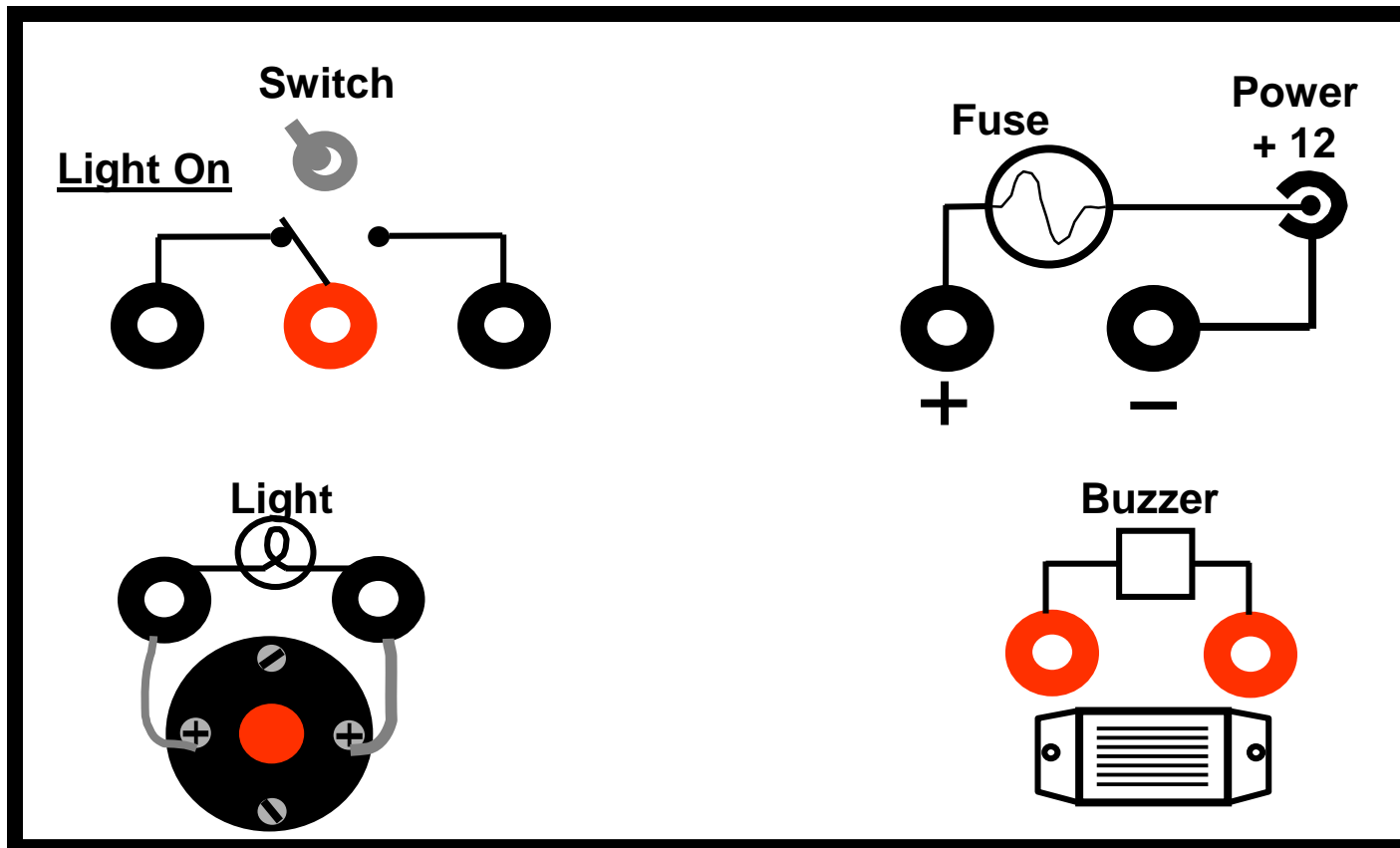
Draw Circuit to Switch Buzzer On / Off - Instructor draws this one first



Instructor \_\_\_\_\_

# Direct Current Test Box

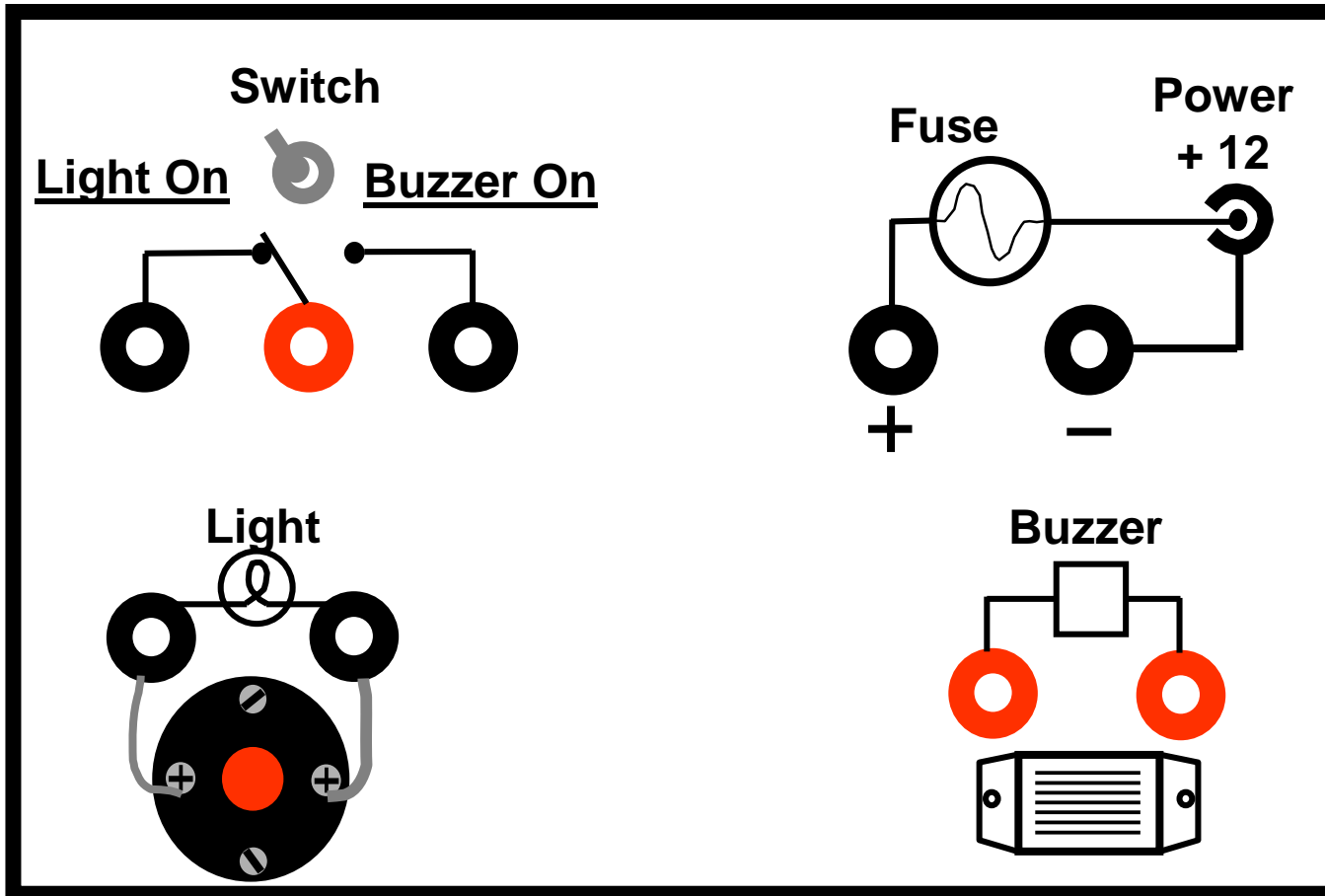
## Draw Circuit to Switch Light On / Off



Instructor \_\_\_\_\_

## Direct Current

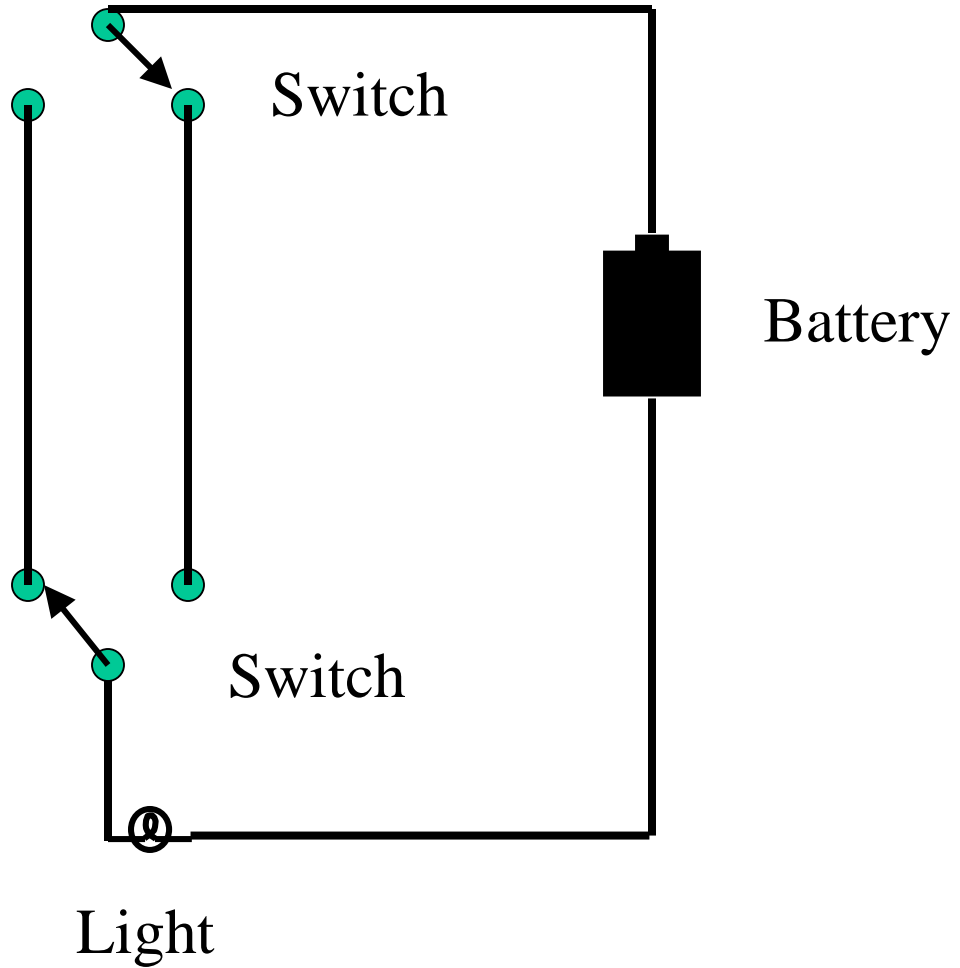
Draw Circuit to Turn Buzzer on in one Direction and Light in other Direction



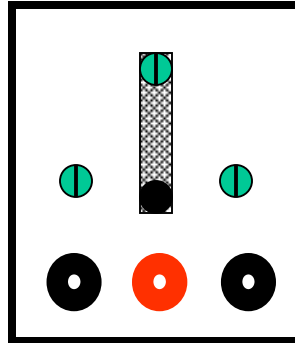
Instructor \_\_\_\_\_

# Two Switches Control One Light

Two different  
Light Switches  
Control one Light

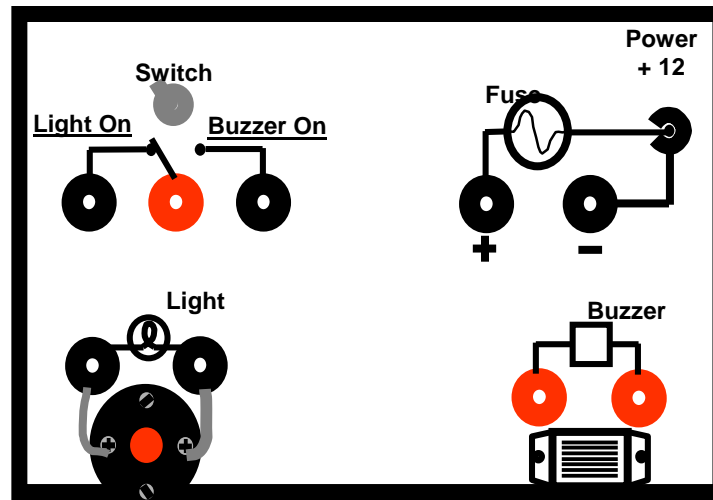


# Draw a Circuit with Two Switch External Switch and On-board Switch



When complete dismantle switch box and put all components back into its box.

Turn Light on



Pass out boxes, connect 4 test circuits  
Students work at own pace

Instructor \_\_\_\_\_